Mercury Toxicity

Mercury

USES	Red ink, thermometers, batteries, fluorescent tubes, electrical appliances, biocides, medications
FORMS	Elemental, metallic Hg (Hg ⁰), divalent inorganic Hg (Hg ²⁺) methylmercury (CH ₃ Hg ⁺ and (CH ₃) ₂)
SPECIES	Monomethyl-Hg is the worst!
NATURE	Soils/rocks, Cinnabar (HgS), food (fish)
HUMAN RELEASE	Combustion of coal and oil, mining/smelting mercury, Cu/Pb smelting, chlor-alkali plants, dental amalgam, fluorescent lamp manufacturing

Mercury - in the body

- Hg²⁺ binds to proteins (affinity for proteins)
- Kidney mistakes protein w/Hg as glutathione (a good guy), and retains it
- Hg is concentrated in the kidneys
- Leads to kidney failure and/or death, can also damage central nervous system

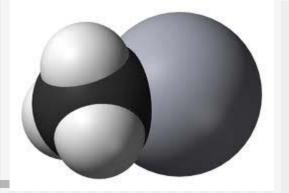
Mercury - in the environment

- Hg⁰ is oxidized and transformed to Methyl-Hg
 @ the water-sediment interface
- Methyl-Hg is the most soluble, persistent, and toxic form of Hg (bio 1/2 life of 70 days in humans - vs 6 days for inorganic Hg)

Factors Affecting Methylation

- Low pH = stimulates CH₃Hg⁺ production, increases total Hg available for methylation (increased binding to particulates and decreased evaporation)
- High DOC (leads to methylmercury in fish even in pristine locations - eg. N.pike in LaGrande reservoir, Quebec)

CH₃Hg⁺

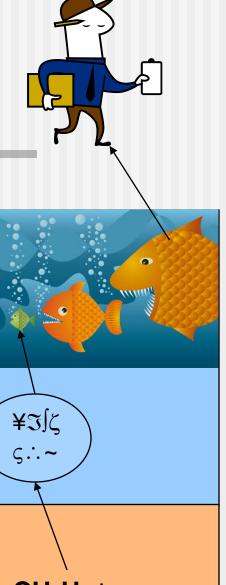


- Affinity for proteins, moves readily across bio membranes, bioaccumulates in fish muscle tissue, and biomagnifies up the food chain
- Fish are tolerant (10x more so than humans) b/c they store Hg in muscles
- Bad news for those that eat the fish!
- Causes decrease egg hatching/growth, and death in fish when levels are very high

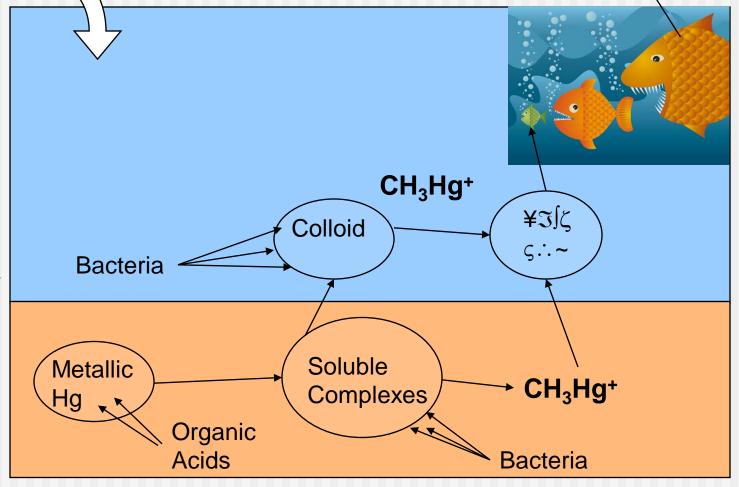


Hg⁰

The Process....



Water Sediment



Effects on Humans

- Impaired hand-eye coordination
- Memory/ Speech loss
- Lowers IQ
- Blurred vision, blindness
- Teratogen (babies born w/twisted limbs)
- Kidney damage
- Cancer
- Death
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Mercury

- Kidneys = affected organs w/ moderate duration exposure to moderate levels
- Brain = receptor in long-term exposure to moderate levels